

Nucleoside Anomeric Radicals *via* 1,5-Translocation: Facile Access to Anomeric Spiro Nucleosides

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Key Words: anomeric radical; 1,5-translocation; nucleoside; 5-*endo-trig* cyclization; anomeric spiro structure

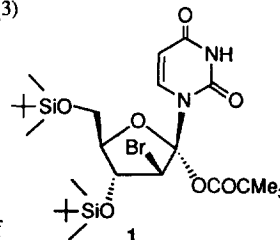
Abstract: A method for generating anomeric radicals of nucleosides was developed based on a 1,5-translocation strategy. A vinyl radical derived from uracil nucleosides having a β,β -dibromovinyl group at the C6 position undergoes the 1,5-translocation to form an anomeric radical, which then cyclizes with the resulting $\text{CH}=\text{CHBr}$ group in a 5-*endo-trig* manner. The whole sequence has disclosed a facile access to anomeric spiro derivatives. A similar reaction of a 6-chloro-9-(β -D-ribofuranosyl)purine derivative was also briefly examined.

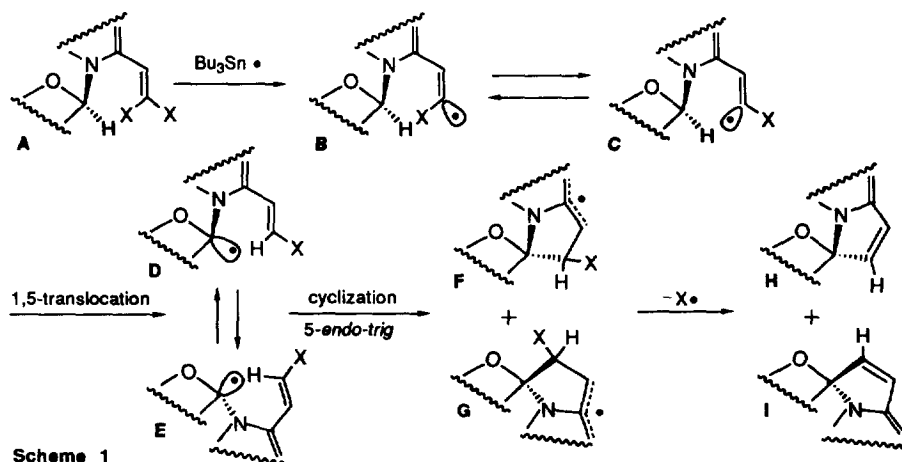
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The finding that anomeric radicals of carbohydrates prefer α -facial reaction,¹⁾ irrespective of the C2-configuration, has offered a great advantage over conventional ionic processes especially in constructing the β -glycosidic linkage.²⁾ These anomeric radicals were generated from carbohydrate derivatives having a suitable C1'-radical source (halogen, SePh, *etc.*). Since the anomeric position of nucleosides is preoccupied by a base moiety, which sometimes acts as a leaving group, there has been no method available for the preparation of such C1'-substituted radical precursors. Therefore, different chemistry has to be developed for generating nucleoside anomeric radicals. Quite recently, we and Chatgililoglu *et al.* independently reported the first example of a nucleoside anomeric radical in which a compound like **1** was used as a precursor.³⁾

In this instance, initially formed C2'-radical undergoes 1,2-translocation with migration of the pivaloyl group to yield an anomeric radical. This communication describes an alternative approach for generating the anomeric radicals, which is based on the 1,5-translocation strategy.⁴⁾

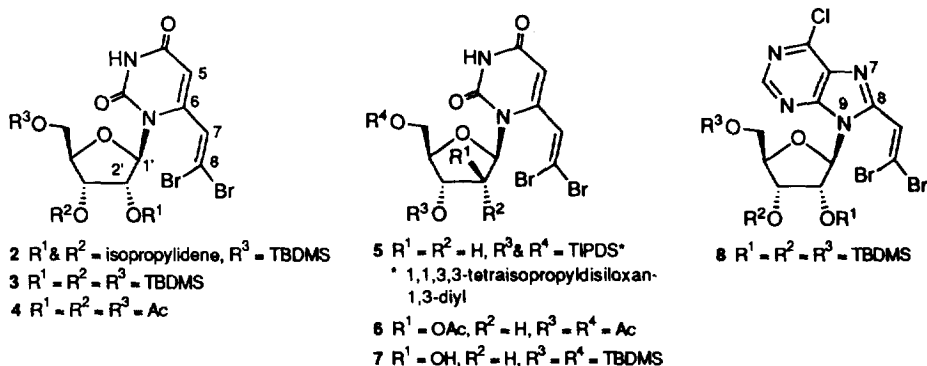
Our whole idea is visualized in Scheme 1. Reaction of a tin radical with **A** having a structure $\text{CH}=\text{CX}_2$ at the C6 position of pyrimidine nucleoside (or C8 of purine nucleoside) gives a vinyl radical **B**, which then forms **C** through radical inversion. Provided that the 1,5-translocation of **C** takes place, a new method of generating an anomeric radical (**D** and **E**) can be developed. Although the subsequent intramolecular reaction leading to anomeric spiro nucleosides **H** and **I** is formally a disfavoured 5-*endo-trig* cyclization,^{5,6)} there would be a good chance in this particular case to effect such a reaction pathway: conjugation of the $\text{CH}=\text{CHX}$ group with an enone structure of uracil ring (or an





imine system of a purine base) would facilitate the formation of incipient radicals F and G which are stabilized by the resonance effect. It should be mentioned that the anomeric β -spironucleoside H serves as a conformationally fixed model of naturally occurring nucleosides.⁷⁾

The β,β -dibromovinyl group was selected as a vinyl radical source, and 2-8 were synthesized by way of lithiation⁸⁾ (for introduction of a formyl group) followed by the Wittig reaction.⁹⁾ Radical reaction of these compounds was carried out by adding a mixture of Bu_3SnH (2.5 equiv) and AIBN (0.5 equiv) over 3 h by a syringe pump to a refluxing benzene solution of each substrate. The results are summarized in Table 1 with only cyclized products being listed. When 2 was used (entry 1), the spiro β -nucleoside 9 was isolated as the major product along with 10 and 11 after HPLC separation (hexane/EtOAc = 1/1-1/2). The 7,8-dihydro-8-phenyl derivative 10 was obtained as a single isomer. The C8-configuration of 10 was assumed to be *R*, based on its ¹H NMR spectrum: a high field shift of H-4' (δ 3.44 ppm), which is located above the phenyl ring, was observed (*cf.* H-4' of 9: δ 4.29 ppm). When this reaction was repeated using Bu_3SnD , no deuterium incorporation was observed in 9. In entry 2, the major product 12 was accompanied with four 7,8-dihydro derivatives (13-16). Similar results are seen in entry 3 wherein 17-21 were obtained (compound 19 consists of (*8S*)- and (*8R*)-isomers). Since the glycosidic conformation of these products is fixed in the *syn*-range, their anomeric



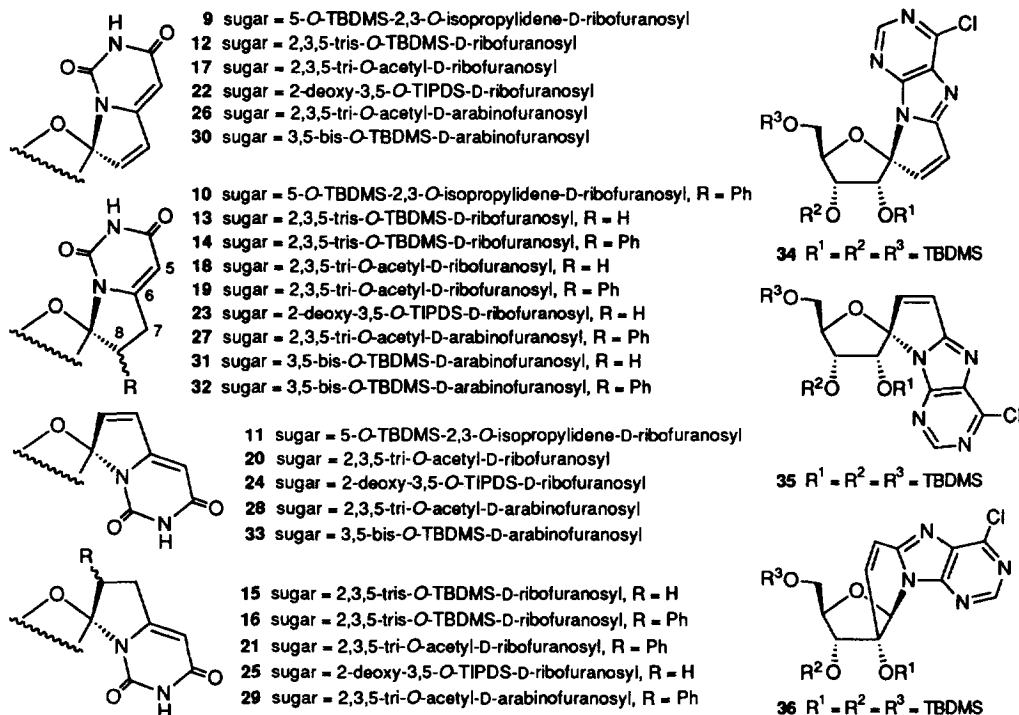


Table 1. Radical-Mediated Cyclization of **2-8**.

Entry	Substrate	Cyclized product (% yield by HPLC isolation)	Combined yield (%)	Ratio of β - and α -isomers (β / α)
1	2	9 (40), 10 (3), 11 (3)	46	ca. 14 / 1
2	3	12 (50), 13 (3), 14 (5), 15 (4), 16 (3)	65	ca. 8 / 1
3	4	17 (40), 18 (1), 19 (13), 20 (7), 21 (1)	62	ca. 7 / 1
4	5	22 (23), 23 (7), 24 (16), 25 (1)	47	ca. 2 / 1
5	6	26 (10), 27 (7), 28 (21), 29 (14)	52	1 / 2
6	7	30 (11), 31 (8), 32 (12), 33 (26)	57	ca. 1 / 1
7	8	34 (18), 35 (14), 36 (5)	37	ca. 1 / 1

stereochemistry is readily assignable based on an anisotropic effect of C2-carbonyl group to H-2' which is observable only in β -isomers (chemical shifts of H-2' in CDCl₃ δ ppm: **17**, 6.44; **18**, 6.39; **19**, 6.19 and 6.40; **20**, 5.53; **21**, 5.28).¹⁰

It is conceivable that ribofuranosyl anomeric radicals prefer α -orientation as depicted in **D**, irrespective of the hydroxy protecting group, presumably due to unfavourable disposition of the bulky base moiety in **E** caused by the 2'-*O*-substituent.¹¹ As shown in entry 4 by the formation of **22-25**, when the α -face is less crowded, a significant decrease in the β/α ratio was an actual issue.

As can be anticipated from steric hindrance of the 2'-*O*-substituent, proportion of β -isomers decreases when the arabinofuranosyl derivatives **6** and **7** were employed (entries 5 and 6). There may be a trend that increase in bulkiness of the 2'-*O*-protecting group results in decrease of the β/α ratio.¹² Finally, reaction of the

ribofuranosylpurine derivative **8** was examined (entry 7). Although we have no clear explanation about the decreased β/α ratio, the observed comparatively lower combined yield would be a reflection of the fact that the vinyl group is bound to a 5-membered imidazole ring and thus the distance between vinyl radical and H-1' is slightly longer than that of uracil cases. This assumption would be supported by the formation of **36** which is assumed to have resulted from 1,6-translocation¹³⁾ of a vinyl radical.

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